# STUDIES ON POLYAZAINDENES SYNTHESIS OF SEVERAL NEW CONDENSED PYRIDAZINE DERIVATIVES 

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Polyazaindenes are purine analogues and as such may act as antimetabolites in purine biochemical reaction ${ }^{1-3}$. In the last few years we were involved in program aimed


IV


II
$\downarrow$

III
RCOOH

VIII

Via, $X=S$
V1b, $x=0$


Scheme 1
at developing of new antischistosomal drugs. Schistosomiosis is one of the most difficult diseases to treat and is a national problem in this country. As pyrazolo[1,5-a]pyrimidines has been reported to have middle activity against schistosoma mosire, we became interested in synthesizing new analogues of these compounds which may be less toxic and more active. As a part of this program, samples of certain azinopyridazines were required. Synthetic approach is represented in Schemes 1 and 2.

$x: 1$



XIV





XVI $1 \mathrm{AC}_{2} \mathrm{O}$

$X \vee 6$
$\mathrm{PhCH}_{2} \mathrm{NHCOCH}_{2} \mathrm{Cl}$

$X \vee a$

Scheme 2

## EXPERIMENTAL

Melting points are uncorrected. ${ }^{1} \mathrm{H}$ NMR spectra were determined in $\left(\mathrm{CD}_{3}\right)_{2} \mathrm{SO}$ on a Varian 60 MHz spectrometer with TMS as internal standard and chemical shifts are expressed as $\delta$ (ppm). IR spectra were obtained on a Pye Unicam SP-1100 spectrophotometer ( KBr ). Analytical data were provided by the Microanalytical centre, Cairo University.

Compound I (m.p. 270 C ), and compound II (m.p. $150^{\circ} \mathrm{C}$ ) were obtained according to reported procedures ${ }^{4}$.

## 3-Hydrazino-4,5,6-triphenylpyridazine III

To a solution of $I I(3.4 \mathrm{~g}, 0.01 \mathrm{~mol})$ in butanol ( 30 ml ) hydrazine hydrate $(0.5 \mathrm{ml})$ was added and the reaction mixture was refluxed for 4 h . After removal of the solvent in vacuo, the residue obtained was crystallized from butanol to give III (yield $60 \%$ ), m.p. 235- $236^{\circ} \mathrm{C}$. For $\mathrm{C}_{22} \mathrm{H}_{18} \mathrm{~N}_{4}$ ( 338.4 ) calculated: $78.08 \% \mathrm{C}, 5 \cdot 36 \% \mathrm{H}$; found: $78.00 \% \mathrm{C}, 5 \cdot 31 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): 3475 , $3350,3200\left(\mathrm{NH}_{2}\right.$ and NH$) ; 1620(-\mathrm{C}=\mathrm{N}-) ; 1560(-\mathrm{C}=\mathrm{C}-) .{ }^{1} \mathrm{H}$ NMR: $5 \cdot 8-6 \cdot 1 \mathrm{~m}, 3 \mathrm{H}$ $\left.\mathrm{NHNH}_{2}\right) ; 7.5-7 \cdot 7 \mathrm{~m} .15 \mathrm{H}(3 \mathrm{Ph})$.

## 6,7,8-Triphenyltetrazolo[1,5-b]pyridazine $V$

Method A: To a suspension of $I I I(3.38 \mathrm{~g}, 0.01 \mathrm{~mol})$ in a mixture of water ( 4 ml ) and acetic acid ( 5 ml ), a solution of sodium nitrite ( 150 mg ) in water ( 2 ml ) is added dropwise at $0^{\circ} \mathrm{C}$. The mixture is left for 2 h at room temperature. The precipitate formed was collected by suction, crystallized from acetic acid and gave $V$ (yield $75 \%$ ), m.p. $225-226^{\circ} \mathrm{C}$. For $\mathrm{C}_{22} \mathrm{H}_{15} \mathrm{~N}_{5}$ (349.4) calculated: $75 \cdot 62 \% \mathrm{C}, 4 \cdot 32 \% \mathrm{H}$; found: $75 \cdot 70 \% \mathrm{C}, 4 \cdot 28 \% \mathrm{H}$. IR spectrum $\left(\mathrm{cm}^{-1}\right): 1640(-\mathrm{N}=$ $=\mathrm{N}=1600(-\mathrm{C}=\mathrm{N}-) .{ }^{1} \mathrm{H}$ NMR: $7 \cdot 4-7 \cdot 8 \mathrm{~m}, 15 \mathrm{H}(3 \mathrm{Ph})$.

Method B: To a solution of $I I(3.4 \mathrm{~g}, 0.01 \mathrm{~mol})$ in dimethylformamide $(20 \mathrm{ml})$ a solution of sodium azide ( 1 g ) in water ( 5 ml ) was added and the reaction mixture was refluxed for 3 h . After removal of the solvent in vacuo an oily residue was obtained, which repeatedly washed with water and recrystallized from acetic acid gave $V$ (yield $45 \%$ ).

### 6.7.8-Triphenyltriazolo[4,3-b]pyridazine-3-thione VIa

A solutions of $I I I(3.38 \mathrm{~g}, 0.01 \mathrm{~mol})$ in carbon disulfide ( 20 ml ) was refluxed for 2 h . After removal of the solvent in vacuo, the residue obtained was crystallized from methanol to give VIa (yield $60 \%$ ), m.p. $260-261^{\circ} \mathrm{C}$. For $\mathrm{C}_{23} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{~S}$ (380.4) calculated: $72 \cdot 61 \% \mathrm{C}, 4 \cdot 24 \% \mathrm{H}$; found: $72 \cdot 65^{\circ} \% \mathrm{C}, 4 \cdot 30 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): $3150(\mathrm{NH}), 1650(\mathrm{C}=\mathrm{N}), 1480(\mathrm{C}=\mathrm{S})$.

## 6,7,8-Triphenyltriazolo[4,3-b]pyridazine-3-one VIb

To a solution of $I I(3.4 \mathrm{~g}, 0.01 \mathrm{~mol})$ in acetonitrile ( 30 ml ) ethyl hydrazinoformate $(1.04 \mathrm{~g}$, 0.01 mol ) was added and the mixture was refluxed for 3 h . After removal of the solvent in vacuo, the residue obtained was crystallized from acetic acid to give VIb (yield $65 \%$ ), m.p. $280-281^{\circ} \mathrm{C}$. For $\mathrm{C}_{23} \mathrm{H}_{16} \mathrm{~N}_{4} \mathrm{O}$ ( $364 \cdot 4$ ) calculated: $75 \cdot 81 \% \mathrm{C}, \mathbf{4 . 4 2 \%} \mathrm{H}$; found: $75 \cdot 79 \% \mathrm{C}, 4 \cdot 41 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): $3200 \mathrm{br}(\mathrm{NH}), 1690(\mathrm{C}=\mathrm{O})$. ${ }^{1} \mathrm{H}$ NMR: $6.4 \mathrm{br}, \mathrm{s}, 1 \mathrm{H}(\mathrm{NH}) ; 7.4-7.7 \mathrm{~m}, 15 \mathrm{H}$ (3 Ph).

### 6.7.8-Triphenyltriazolo[4,3-b]pyridazine VIIa

A solution of $I I I(3.38 \mathrm{~g}, 0.01 \mathrm{~mol})$ in formic acid ( 30 ml ) was refluxed for 4 h . After removal of the solvent in vacuo, the residue obtained was crystallized from benzene to give VIIa (yield $65^{\circ} \%$, m.p. $255-256^{\circ}$ C. For $\mathrm{C}_{23} \mathrm{H}_{16} \mathrm{~N}_{4}(348 \cdot 4)$ calculated: $79 \cdot 29 \% \mathrm{C}, 4 \cdot 62 \% \mathrm{H}$; found: $79 \cdot 30 \% \mathrm{C}$, $4.58^{\circ}{ }_{\circ}^{\prime} \mathrm{H} .{ }^{1} \mathrm{H}$ NMR: $6.2 \mathrm{~s}, 1 \mathrm{H}$ (H7); 7.4-7.7 m, 15 H (3 Ph).

3-Methyl-6,7,8-triphenyltriazolo[3,4-b]pyridazine VIIb
A solution of $I I I(3.38 \mathrm{~g}, 0.01 \mathrm{~mol})$ in acetic acid $(30 \mathrm{ml})$ was refluxed for 4 h . After removal of the solvents in vacuo, the residue obtained was crystallized from benzene to give VIIb (yield $75 \%$ ), m.p. $265-266^{\circ}$ C. For $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{~N}_{4}(362 \cdot 4)$ calculated: $79 \cdot 54 \% \mathrm{C}, 5 \cdot 00 \% \mathrm{H}$; found: $79 \cdot 60 \% \mathrm{C}$, $4.91 \%$ H. ${ }^{1} \mathrm{H}$ NMR: $2 \cdot 3 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{CH}_{3}\right) ; 7 \cdot 5-7 \cdot 8 \mathrm{~m}, 15 \mathrm{H}$ (3 Ph).

## 4,5,6-Triphenyl-2H-pyridazine-3-thione VIII

Method A: To a solution of $I(3.24 \mathrm{~g}, 0.01 \mathrm{~mol})$ in pyridine ( 20 ml ) was added phosphorus pentasulfide $(4.4 \mathrm{~g})$. The reaction mixture was heated under reflux for 4 h . The resulting solution was poured into crushed ice ( 400 g ) containing $\mathrm{HCl}(70 \mathrm{ml})$. The yellow solid product obtained was filtered off and crystallized from acetic acid to give VIII (yield $55.5 \%$ ), m.p. $250-251^{\circ} \mathrm{C}$. For $\mathrm{C}_{22} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{~S}(340 \cdot 4)$ calculated: $77 \cdot 64 \% \mathrm{C}, 4 \cdot 11 \% \mathrm{H}$; found: $77 \cdot 70 \% \mathrm{C}, 4 \cdot 10 \% \mathrm{H} .{ }^{1} \mathrm{H}$ NMR: $3.25 \mathrm{~s}, 1 \mathrm{H}(\mathrm{SH}) ; 7.78 \mathrm{~m}, 15 \mathrm{H}$ ( 3 Ph ).

Method B: Thiourea ( $1.14 \mathrm{~g}, 15 \mathrm{mmol}$ ) was added to a solution of $I I(3.4 \mathrm{~g}, 10 \mathrm{mmol})$ in 25 ml of absolute ethanol and the reaction mixture was stirred at $20^{\circ} \mathrm{C}$ for 3 h . The light yellow deposit thus obtained was filtered and treated with 2.5 m sodium hydroxide ( 14 ml ) and refluxed for 1 h . The reaction mixture was concentrated to half of its volume, cooled in an ice bath and acidified with hydrochloric acid ( pH 2 ). The yellow product thus obtained was filtered and recrystallized from acetic acid to give VIII (yield $70 \%$ ).

## 3-Benzylmethylthio-4,5,6-triphenylpyridazine $X$

To a solution of $V I I I(3.48 \mathrm{~g}, 0.1 \mathrm{~mol})$ and sodium ethoxide ( 0.35 g of sodium) in ethyl alcohol ( 30 ml ), freshly prepared phenacylbromide $(2 \mathrm{~g}, 0.01 \mathrm{~mol}$ ) was added. The reaction mixture was refluxed for 2 h . A crystalline matter was separated and recrystallized from ethanol to give $X$ (yield $62 \%$ ), m.p. $215-216^{\circ} \mathrm{C}$. For $\mathrm{C}_{30} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{OS}$ (458.6) calculated: $78.57 \% \mathrm{C}, 4 \cdot 83 \% \mathrm{H}$; found: $78.50 \% \mathrm{C}, 4 \cdot 79 \% \mathrm{H}$. IR spectrum $\left(\mathrm{cm}^{-1}\right): 1690(\mathrm{C}=\mathrm{O}), 1620(\mathrm{C}=\mathrm{N}) .{ }^{1} \mathrm{H}$ NMR: 3.8 s , $2 \mathrm{H}\left(\mathrm{CH}_{2}\right) ; 7 \cdot 7-7 \cdot 9 \mathrm{~m}, 20 \mathrm{H}(4 \mathrm{Ph})$.

3-Methylthio-4,5,6-triphenylpyridazine $X I$
To a solution of VIII $(3.48 \mathrm{~g}, 0.01 \mathrm{~mol})$ in ethanol $(30 \mathrm{ml})$ was added aqueous sodium hydroxide $\left(0.5 \mathrm{~mol} 1^{-1}, 10 \mathrm{ml}\right)$ and methyl iodide $(0.4 \mathrm{~g})$. The reaction mixture was stirred for 2 h at $20^{\circ} \mathrm{C}$. The light yellow deposit thus obtained was extracted with dichloromethane. The dichloromethane was removed in vacuo and the residue was crystallized from ethanol to give $X I$ (yield $75 \%$ ), m.p. $230-231^{\circ} \mathrm{C}$. For $\mathrm{C}_{23} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{~S}(354 \cdot 5)$ calculated: $77 \cdot 93 \% \mathrm{C}, 5 \cdot 11 \% \mathrm{H}$; found: $77 \cdot 87 \% \mathrm{C}, 5 \cdot 09 \% \mathrm{H}$. IR spectrum $\left(\mathrm{cm}^{-1}\right): 1580(\mathrm{C}=\mathrm{N}), 1500(\mathrm{C}=\mathrm{C}) .{ }^{1} \mathrm{H}$ NMR: $3.52 \mathrm{~s}, 3 \mathrm{H}\left(\mathrm{SCH}_{3}\right) ; 7 \cdot 5-7 \cdot 3 \mathrm{~m}$, 15 H (3 Ph).

Reaction of XI with hydrazinehydrate: To a solution of $X I(3.5 \mathrm{~g}, 0.01 \mathrm{~mol})$ in ethanol ( 30 ml ), hydrazine hydrate $(0.5 \mathrm{ml})$ was added and the reaction mixture was refluxed for 6 h . After removal of the solvent in vacuo the residue obtained was crystallized from butanol to give III (yield $80 \%$ ) which was identified by a comparison of its physical properties with that from the former experiment.

## 3-Benzoylmethyl-4,5,6-triphenylpyridazine XII

A solution of $X(1 \mathrm{~g}, 2 \mathrm{mmol})$ in pyridine ( 15 ml ) was refluxed for 30 min . The reaction mixture was poured into cold diluted HCl . A solid matter was separated and crystallized from methanol
to give XII (yield $55 \%$ ), m.p. $260-261^{\circ} \mathrm{C}$. For $\mathrm{C}_{30} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}(427 \cdot 5)$ calculated: $84 \cdot 28 \% \mathrm{C}$, $5.42 \%$ H; found: $84.21 \% \mathrm{C}, 5.45 \%$ H. IR spectrum $\left(\mathrm{cm}^{-1}\right): 1685(\mathrm{C}=\mathrm{O}), 1550(\mathrm{C}=\mathrm{C}) .{ }^{1} \mathrm{H}$ NMR: $3.2 \mathrm{~s}, 2 \mathrm{H}\left(\mathrm{CH}_{2}\right) ; 7 \cdot 4-7 \cdot 8 \mathrm{~m}, 20 \mathrm{H}(4 \mathrm{Ph})$.

## 3-Thiobenzoylmethyl-4,5,6-triphenylpyridazine XIV

To a solution of $X I I(2 \mathrm{~g}, 4 \mathrm{mmol})$ in dry xylene ( 50 ml ), phosphorous pentasulfide $(3 \cdot 1 \mathrm{~g}$, 0.014 mol ) was added. The reaction mixture was refluxed for 2 h , filtered hot, the filtrate was concentrated, orange crystals separated and recrystallized from benzene to give $X I V$ (yield $45 \%$ ), m.p. $280-281^{\circ} \mathrm{C}$. For $\mathrm{C}_{30} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{~S}(442 \cdot 6)$ calculated: $81.41 \% \mathrm{C}, 5.01 \% \mathrm{H}$; found: $81.37 \% \mathrm{C}$, $4.92 \%$ H. IR spectrum $\left(\mathrm{cm}^{-1}\right): 1560(\mathrm{C}=\mathrm{C}), 1475(\mathrm{C}=\mathrm{S}) .{ }^{1} \mathrm{H}$ NMR: $3.1 \mathrm{~s}, 2 \mathrm{H}\left(\mathrm{CH}_{2}\right) ; 7.6$ to $7.9 \mathrm{~m}, 20 \mathrm{H}(4 \mathrm{Ph})$.

## 3-Benzylaminocarbonylmethylthio-4,5,6-triphenylpyridazine $X V a$

A mixture of VIII ( $3.48 \mathrm{~g}, 0.01 \mathrm{~mol}$ ), potassium carbonate ( 1 g ), and monochlorobenzyl amide $(1.83 \mathrm{~g}, 0.01 \mathrm{~mol})$ was refluxed in dry acetone for 12 h . After removal of the solvent in vacuo, water ( 20 ml ) was added, and reaction mixture extracted with ether ( $3 \times 50 \mathrm{ml}$ ). The extracts were washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated. The residue was crystallized from methanol to give $X V a$ (yield $62 \%$ ), m.p. $198-199^{\circ}$ C. For $\mathrm{C}_{31} \mathrm{H}_{25} \mathrm{~N}_{3} \mathrm{OS}$ (487.6) calculated: $76.35 \% \mathrm{C}, 5 \cdot 16 \% \mathrm{H}$; found: $76.40 \% \mathrm{C}, 5 \cdot 09 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): $3150(\mathrm{br}, \mathrm{NH}), 1700$ ( $\mathrm{C}=\mathrm{O}$ ) .

## 3-Carboxymethylthio-4,5,6-triphenylpyridazine $X V b$

To a solution of $V I I I(3.48 \mathrm{~g}, 0.01 \mathrm{~mol})$ in acetic acid ( 30 ml ), chloroacetic acid $(0.94 \mathrm{~g}, 0.01 \mathrm{~mol})$, and sodium acetate ( 3 g ) were added. The reaction mixture was refluxed for 5 h . After removal of the solvent in vacuo, an oily residue was obtained which was repeatedly washed with water and recrystallized from methanol to give $X V b$ (yield $60 \cdot 6 \%$ ), m.p. $155-156^{\circ} \mathrm{C}$. For $\mathrm{C}_{24} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}$ (398.5) calculated: $72.34 \% \mathrm{C}, 4.55 \% \mathrm{H}$; found: $72 \cdot 28 \% \mathrm{C}, 4.51 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): 3100 $(\mathrm{OH}), 1725(\mathrm{C}=\mathrm{O}) .{ }^{1} \mathrm{H}$ NMR: $3.9 \mathrm{~s}, 2 \mathrm{H}\left(\mathrm{CH}_{2}\right) ; 7 \cdot 4-7 \cdot 6 \mathrm{~m}, 15 \mathrm{H}(3 \mathrm{Ph}) ; 12 \cdot 1 \mathrm{~s}, 1 \mathrm{H}(\mathrm{OH})$.

## 6,7,8-Tripheayl-1,3-thiaz slo 2,3-b pyridazinium-3-one $X V I$

Acetic anhydride ( 10 ml ) was added at room temperature to a solution of $X V b(1 \mathrm{~g}, 2 \mathrm{mmol})$ in pyridine ( 10 ml ). The reaction mixture was immediately warmed on a steam-bath for 5 min . The colour of the solution quickly changed from light to deep yellow and finally to redish brown when the reaction was virtually complete. The mixture was immediately cooled down to room temperature and the excess of acetic anhydride decomposed with $95 \%$ ethanol ( 10 ml ). The pale yellow mesoionic $X V I$ separated on keeping the reaction mixture for 2 h in the cold-bath, was crystallized from ethyl acetate (yield $13 \%$ ), m.p. $95-96^{\circ}$ C. For $\mathrm{C}_{24} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{OS}(381 \cdot 5)$ calculated: $75.56 \% \mathrm{C}, 4.49 \% \mathrm{H}$; found: $75.51 \% \mathrm{C}, 4.41 \% \mathrm{H}$. IR spectrum ( $\mathrm{cm}^{-1}$ ): $1670(\mathrm{C}=\mathrm{O}), 1600$ ( $\mathrm{C}=\mathrm{C}$ ).

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